Chemistry 20: Non, Total, Net ionic reactions

1. **Numerical response question**

|  |  |  |  |
| --- | --- | --- | --- |
|  |  |  |  |

Left justify your answer in the boxes provided

|  |
| --- |
| Consider the following total ionic equation.2K+(aq) + CO32-(aq) + Ca2+(aq) + 2Cl-(aq) 🡪 2K+(aq) +2Cl-(aq) + CaCO3(s)Identify the spectator ion(s) in this reaction from the list below:1. K+(aq)
2. CO32-(aq)
3. Ca2+(aq)
4. Cl-(aq)
5. CaCO3(s)
 |

Solutions:

1. 14